Study of the Effects of Chayotte (*Sechium Edule*) Extracts on the Plasmid pUC. 9.1 DNA

Gláucio F. Diré¹, Maria L. Gomes¹, Elaine A. C. Lima¹, R. L. Jales³, M. Castro Faria¹, Mario Bernardo-Filho^{1,2}

(1. Universidade do Estado do Rio de Janeiro, Instituto de Biologia Roberto Alcantara Gomes, Departamento de Biofísica e Biometria;

2. Instituto Nacional do Câncer, Centro de Pesquisa Básica, Praça Cruz Vermelha;

3. Universidade Federal do Rio Grande do Norte, Departamento de Farmácia, Natal, Rio Grande do Norte, Brazil)

Abstract: Stannous chloride (SnCl₂) is employed as a reducing agent to obtain Technetium-99m-labelled radiopharmaceuticals in nuclear medicine kits, being inject endovenously in humans. Toxic effects of these kits were not studied, thus making it important to evaluate their impact in humans. The use of natural extracts as medicines is growing around the world. The chayotte (Sechium edule) is a subtropical vegetable with potent diuretic action. It is used in folk medicine due its hypotensor effect. In this study, plasmid deoxyribonucleic acid (DNA) was exposed to chayotte extracts (macerated and decoct) (0.1 g.mL⁻¹) in presence of stannous chloride (SnCl₂). Samples of the plasmid DNA were analyzed through agarose gel electrophoresis. The results show that the chayotte extracts were capable of damaging the DNA in the presence and in the absent of SnCl₂ [Nature and Science. 2004;2(3):48-54].

Key words: plasmidial DNA, chayotte, nuclear medicine, stannous chloride.

1 Introduction

Stannous chloride can cause skin and mucosal irritation in humans, and when this salt is injected into laboratory animals, it can produce stimulation and subsequent depression of the central nervous system (Gleason *et al.*, 1969). It has been suggested that SnCl₂ is a powerful genotoxic (McLean *et al.*, 1983; Oliver and Marzin, 1987), mutagenic (Singh, 1983; Triphaty *et al.*, 1990) and carcinogenic (Ashby and Tennant, 1991) compound. In nuclear medicine, SnCl₂ has been employed in scintigraphic test as Technetium–99m (99mTc) reducing agent. Besides the use of SnCl₂ in nuclear medicine, this salt is also used in dentistry (dentifricies) (Hallas and Cooney, 1981; McLean *et al.*, 1983; Rader, 1991; White, 1995; Budavery, 1996).

There are other sources of $SnCl_2$ to which human beings are exposed to such as from environmental contamination by biocide preparations containing organic compound dimethyl stannous chloride [SnCl₂ (CH₃)₂] (Hallas and Cooney, 1981). It is hypothesized that the toxicity of SnCl₂ might be mediated by generation of reactive oxygen species (ROS) through the reaction: $\operatorname{Sn}^{2+} + \operatorname{O}_2 + 2\operatorname{H}^+ \rightarrow \operatorname{Sn}^{4+} + \operatorname{H}_2\operatorname{O}_2$. The generation hydrogen peroxide undergoes by Fenton reaction to generate [•]OH as follows: $Fe^{2+} + H_2O_2 \rightarrow$ OH⁻ + [•]OH (4). It was also described that SnCl₂ mediates single strand breaks in plasmid DNA through ROS formation in a dose-dependent manner (Dantas et al., 1996). In addition, the mutagenic potentiality of SnCl₂ was identified by *supF* gene mapping (Cabral *et* al., 1998). It was also determined that Escherichia coli (E. coli) strains proficient in DNA repair mechanisms were more resistant to SnCl₂ treatment than deficient ones, suggesting that inactivation was due to DNA damage (Aherne and O'Brien, 1999). Biological effects of metals have been reported: (i) transition metals catalyze free radical production that can be related to aging processes and neurodegenerative diseases such as Alzheimer's diseases, Parkinson's disease, and others (Stohs and Bagchi, 1995); (ii) the association between human diseases and metal ions metabolism can be also demonstrated by Huntington's disease and amyotrophic lateral sclerosis; (iii)

neurotoxic properties of aluminum and its etiopathogenetic role in Alzheimer'a disease are still controversially argued and need further elucidation (Gutteridge et al., 1984); (iv) abnormal copper metabolism detected in the brain tissue of Wilson's and Menke's patients remains to be fully explained (Gutteridge et al., 1985); (v) substantial levels of zinc are found in the hippocampus and its brain deficiency can be the cause of several pathological events (Bettger and O'Dell, 1981), however; (vi) magnesium is successfully used for the treatment of migraine (Stohs and Bagchi, 1995). In addition, the molecular mechanisms underlying neurotoxicity associated with mercury, tin and manganese, also need further investigation. Tin is a heavy metal which has long been regarded as a contaminant of the environment (Wood, 1974). One of its inorganic salts, stannous chloride (SnCl₂), has been widely used in daily human life, to conserve soft drinks, in food manufacturing, as a result of processing and packaging. Studies on the biological effects of SnCl₂ revealed that it can generate reactive oxygen species (ROS) and breaks in deoxyribonucleic acid (DNA) (Caldeira-de-Araújo et al., 1996) and induces lethality in E. coli, whose damage recovery depends on RecA-mediated repair (Bernardo-Filho et al., 1994b). Medicinal plants are mainly complex products with several components with different chemical and pharmacological characteristics (Moro and Basile, 2000). In addition, many of these products are also sold as dietary supplement, but, scientific information about their safe and effective use is hard to find because limited toxicological data are available on herbal remedies and support of rigorous clinical studies is lacking (Capasso et al., 2000). The use of natural products as medicines has been growing in the entire world. Because of this fact, many studies with natural products are being developed, and new drugs for treatments of diseases are being discovered. In the literature, the medicinal action mechanism of several plants has been described and different compounds, with various properties, have been isolated from the crude extracts (Leite et al., 1986; Sallé, 1996). Sechium edule (chayotte), a subtropical vegetable with potent diuretic action, is a cucurbitaceus species which is used as food or as medication in popular medicine. It was reported a case of severe hypokalemia pregnancy and that a chayotte preparation was implicated, as the potassium level returned to normal,

without recurrence of hypokalemia, once the ingestion of this vegetable stopped (Jensen and Lai, 1986). Gordon et al. (2000) described the hypotensor effect of the chayotte. Diré et al. (2002) have shown that the extracts of chayotte (macerated and infusion) were capable of altering the labeling of blood elements with technetium-99m (99mTc) in an in vivo study. In other research Diré et al. (2001) have demonstrated that a chayotte extract (macerated) was able to alter the biodistribution of 99mTc as sodium pertechnetate (NaTcO₄) as well as the shape of red blood cells through a qualitative analysis. The effect of stannous ion has been abolished by extracts of some medicinal plants (Reiniger et al., 1999; Melo et al, 2001; Lima et al., 2002, Silva et al., 2002). Bernardo et al. (2002) described that the rutin, a compound isolated from Ruta graveolens, was not capable of damaging DNA, protecting DNA from the SnCl₂ redox action and inactivating the Escherichia coli (E. coli AB1157) culture.

ROS are generated during a variety of cellular events with beneficial as well as deleterious effects to the organism (Halliwell, 1994). Some plant extracts may increase the effects of the deleterious actions of ROS (Lima *et al.*, 2001). In the present study, we have evaluated the influence of a chayotte extracts on the topology on gel electrophoretic of plasmid DNA submitted to $SnCl_2$.

2 Material and Methods

2.1 Characterization of the chayotte sample

Chayotte was purchased from a local market in Rio de Janeiro city, RJ, Brazil. To prepare the extract, 50 g of the skin of chayotte were mixtured with 500 ml of water in an electric extractor. This preparation was filtered and this extract was considered 100%.

The presence of toxic compounds was evaluated and we did not find them in the extracts of chayotte used in our experiments. The method to verify the presence of these toxic products is based on inhibition of acethylcholinesterase in the presence of the pesticides (Cunha Bastos *et al.*, 1991). In this method, brain acethylcholinestarase is utilized as an *in vitro* detector of organophosphorus and carbamate insecticides. Briefly, a preparation of a cethylcholinesterase was obtained after extraction of a rat brain microsomal fraction with Triton X-100 and was incubated with the extract of cauliflower. Enzyme assay was performed by a potentiometric method based on the formation of acetic acid in the incubation mixture (preparation of acethylcholinesterase and extract of chayotte).

2.2 Nucleic acid manipulations

Plasmids were diluted, dispensed into eppendorf tubes (200 ng/tube) and incubated with 200 µg.mL⁻¹ of SnCl₂. To evaluate the influence of the extract of the chayotte in DNA breakage, a concentration on a par with 0.1 g.mL⁻¹ was used. In all cases, reaction mixtures were incubated at 37°C for 40 min. The analysis of the single breaks (SSB) formation was performed using 0.8% agarose gel electrophoresis in order to separate the conformations of plasmid DNA: form I supercoiled native conformation and form II open circle resulting from SSB. Aliquots from each sample (10 µL) were mixed to 2 µL of 6x concentrated loading buffer (0.25% xylene cyanol FF; 0.25% bromofenol blue; 30% glycerol), and applied in a horizontal gel electrophoresis chamber in Tris acetate-EDTA buffer at pH 8.0. After electrophoresis, the gel was stained with ethidium bromide (0.5 $\mu g.mL^{-1}$)

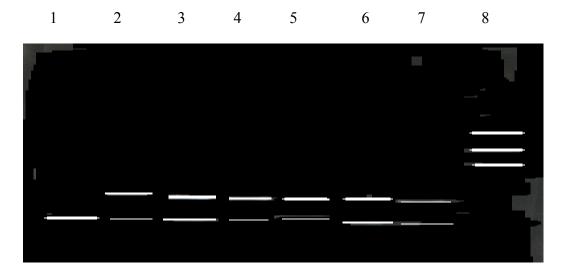
and the DNA bands were visualized by fluorescence in an ultraviolet (UV) transiluminator system. Permanent records were performed using a polaroid MP- 4^+ system.

3 Results

Figure 1 shows electrophoresis of pUC. 9.1 plasmid with SnCl₂ and/or the extract of macerated extract, and the electrophoretic mobility of plasmidial DNA in various experimental conditions (macerated extract) in agarose gel.

Figure 2 showselectrophoresis pUC. 9.1 plasmid with $SnCl_2$ and/or the extract of decoct extract (heated till 100°C) and electrophoretic mobility of plasmidial DNA in various experimental conditions (decoct extract) in agarose gel.

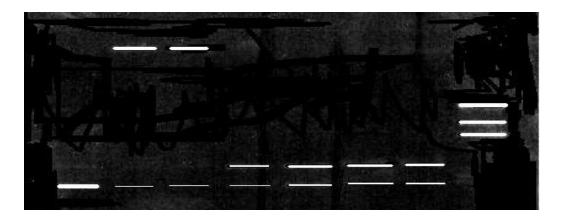
Figure 3 shows electrophoresis of pUC. 9.1 plasmid with $SnCl_2$ and/or the extract of decoct extract (heated till 50°C). and electrophoretic mobility of plasmidial DNA in various experimental conditions (heated extract) in agarose gel.



Column 1: (Control: DNA + water); Column 2: (Chayotte100%); Column 3: (SnCl₂ 200 μ g.mL⁻¹); Column 4: (SnCl₂- 200 μ g.mL⁻¹ + Chayotte 100%); Column 5: (oxidized Chayotte-10min); Column 6: (oxidized SnCl₂- 10min); Column 7 (oxidized Chayotte+ SnCl₂- 10min) ; Column 8 (marker λ hind III). Photos of the gels were scanned.

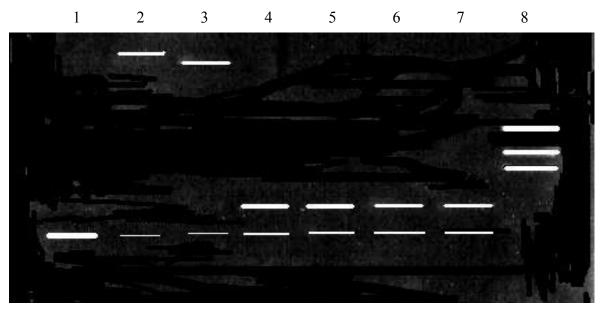
Figure 1. Electrophoresis of pUC. 9.1 plasmid with SnCl₂ and/or the extract of macerated extract and electrophoretic mobility of plasmidial DNA in various experimental conditions (macerated extract) in agarose gel

1 2 3 4 5 6 7 8



Column 1: (Control: DNA + water); Column 2: (Chayotte100%); Column 3: (SnCl₂ 200 μ g.mL⁻¹); Column 4: (SnCl₂- 200 μ g.mL⁻¹ + Chayotte 100%); Column 5: (oxidized Chayotte-10min); Column 6: (oxidized SnCl₂- 10min); Column 7 (oxidized Chayotte+ SnCl₂- 10min) ; Column 8 (marker λ hind III). Photos of the gels were scanned.

Figure 2. Electrophoresis of pUC. 9.1 plasmid with SnCl₂ and/or the extract of decoct extract (heated till 100°C) and electrophoretic mobility of plasmidial DNA in various experimental conditions (decoct extract) in agarose gel



Column 1: (Control: DNA + water); Column 2: (Chayotte100%); Column 3: (SnCl₂ 200 μ g.mL⁻¹); Column 4: (SnCl₂- 200 μ g.mL⁻¹ + Chayotte 100%); Column 5: (oxidized Chayotte-10min); Column 6: (oxidized SnCl₂- 10min); Column 7 (oxidized Chayotte+ SnCl₂- 10min); Column 8: (marker λ hind III). Photos of the gels were scanned.

Figure 3. Electrophoresis of pUC. 9.1 plasmid with SnCl₂ and/or the extract of decoct extract (heated till 50°C). and electrophoretic mobility of plasmidial DNA in various experimental conditions (heated extract) in agarose gel

4 Discussion

Much effort has focused on the identification of phytochemicals in plants, which exert biological effects. The knowledge of these effects are worthwhile and can help to prevent possible undesirable actions of crude extracts and/or purified substances isolated from various plants. Moreover, many times the results reported in the literature are controversies. This fact could be explained by (i) various experimental conditions and models, (ii) the characteristic and the concentration of the used material (crude extract, isolated fraction, purified substance or heated extract), and (iii) the specific condition of the growth of the studied plant. Many biological effects have been associated with the flavonoids and other antioxidant molecules (Webster et al., 1996; Aherne et al., 1999). Reactive oxygen species (ROS) have been implicated as the primary destructive intermediates in a wide range of environmental conditions as well as in an increasing number of humans disorders (mutagenesis, apoptosis, aging) (Hladik et al., 1987). SnCl₂ has been used as a reducing agent (Bernardo-Filho et al., 1994; Caldeira-de-Araujo et al., 1996) in medical procedures.

Cytotoxic and genotoxic SnCl₂-induced damage were demonstrated in *E. coli* and the effects appeared to be mediated by ROS (Caldeira-de-Araújo *et al.*, 1996; Dantas *et al.*, 1996; Felzenszwalb *et al.*, 1998; Dantas *et al.*, 1999; Reiniger *et al.*, 1999).

The treatment of the E. coli strains AB1157 with SnCl₂ in presence of *Peumus boldus* (Reiniger et al., 1999), Cymbopogon citratus, Maytenus ilicifolia, Baccharis genistelloides (Melo et al., 2001), Rutin (Bernardo et al., 2002) and with Brassica oleracea L. var botrytis (Lima et al., 2002) induced protection of cells against the citotoxic-SnCl₂ effects. It could probably be due to oxidant properties of these extracts. However, the intensity of the protective effect against the SnCl₂ effects was dependent on the considered extract. Like observed in the study of the cauliflower extract (Lima et al., 2002) the extract of chayotte was capable of inducing lesion of break type in the plasmid pUC 9.1 DNA (Lima et al., 2001). In comparison with the Bernardo et al., 2002, study, it was verified that the rutin different of the chayotte extracts have not induced lesion in the DNA molecule. In this study the extracts of chayotte (macerated and

decoct) in all concentrations tested have induced lesions in the DNA molecule. In the analysis of Peumus boldus extract was noticed that it has reduced or abolished the effect of SnCl₂ although the lesive effect of boldine was observed when the highest concentration of this substance was used in the presence of the reducing agent despite boldine alone has not been capable of inducing alterations in the DNA. Silva et al., 2002, working with different cultures of E. coli strains also have shown a reduction of the lethal effect induced by SnCl₂ on the survival of the cultures in the presence of C. citratus, B. genistelloides, M. ilicifolia and P. boldus. In the present work it was verified that the heated extracts have induced more breaks in the DNA molecule in comparison with the crude one.

5 Conclusion

In general we can speculate that the extracts of chayotte were capable of inducing damages in pUC. 9.1 DNA molecules and considering the heated extracts it was observed that these extracts have induce more breaks than the macerated one.

Acknowledgments

This research was supported by CNPq, CAPES, FAPERJ and UERJ.

Correspondence to:

Mario Bernardo-Filho, Ph.D.

Universidade do Estado do Rio de Janeiro

Instituto de Biologia Roberto Alcantara Gomes

Departamento de Biofísica e Biometria

Av. 28 de setembro, 87, Rio de Janeiro, RJ, BRASIL 20551-030

Fax number: 55 21 2543532 E-mail: gdire@hotmail.com

References

- Ashby J, d Tennant RW. Definitive relationships among chemical and mutagenicity for 301 chemicals tested by the U.S. NTP. Mutat Res 1991;257:229-306.
- [2] Aherne SA, O'Brien MN. Protection by the flavonoids myricetin, quercetin and rutin against hydrogen peroxidase-induced DNA damage in Caco-2 Nd Hep G2 cells. Nutr Cancer 1999;34:160-6.
- [3] Bernardo-Filho M, Cunha MC, Valsa JO, Araujo AC, Silva FCP, Fonseca AS. Evaluation of potential genotoxicity of

stannous chloride: inactivation, filamentation and lysogenic induction of *Escherichia coli*. Food Chem Toxicol 1994b; 32:477-9.

- [4] Bernardo LC, Oliveira MBN, Silva CR, Dantas FJS, Mattos JCP, Caldeira-de-Araújo A, Moura RS, Bernardo-Filho M. Biological effects of rutin on the survival of Escherichia coli AB1157 and on the electrophoretic mobility of plasmid pUC 9.1 DNA. Cell Mol Biol 2002;48:517-20.
- [5] Bettger WJ, O'Dell BLA. Critical physiological role of zinc in structure and function of biomembranes. Life Sci 1981;28:1425-38.
- [6] Budavery S. The Merck Index. Merck. White-house Station, NJ. 1996:1500-1.
- [7] Cabral REC, leitão AC, Lage C, Caldeira-de-Araújo A, Bernardo-Filho M, Dantas FJS, Cabral-Neto JB. Mutational potentiality of stannous chloride: an important reducing agent in the Tc-99m radiopharmaceuticals. Mutat Res 1998;408:129-35.
- [8] Caldeira-de-Araújo A, Dantas FJS, Moraes MO, Felzenszwalb I, Bernardo-Filho M. Stannous chloride participate in the generation of reactive oxygen species. J Braz Assoc Adv Sci 1996;48:109-13.
- [9] Capasso R, Izzo AA, Pinto L, Bifulco T, Vitobello C, Mascolo M. Phytotherapy and quality of herbal medicines. Fitoterapia 2000;71:S58-S65.
- [10] Cunha Bastos VLF, Cunha Bastos JF, Lima JS. Brain acethylcholinesterase as an *in vitro* detector of organophosphorus and carbamate insectides in the water. Water Res 1991;7:835-40.
- [11] Dantas FJS, Moraes MO, De Mattos JCP, Bezerra RJAC, Carvalho EF, Bernardo-Filho M, Caldeira-de-Araújo A. Stannous chloride mediates single strand breaks in plasmid DNA through reactive oxygen species formation. Toxicol Lett 1999;110:129-36.
- [12] Diré G, Lima E, Mattos D, Oliveira MB, Pereira MJ, Moreno S, Freitas R, Gomes ML, Bernardo-Filho M. Effect of chayotte (*Sechium edule*) extract on the biodistribution of technetium-99m and on the morphometry of red blood cells. J Labelled Cpd Radiopharm 2001;44:648-50.
- [13] Dantas FJS, Moraes MO, Carvalho EF, Valsa JO, Bernardo-Filho M, Caldeira-de-Araújo A. Lethality induced by stannous chloride on Escherichia coli AB1157: participation of reactive oxygen species. Food Chem Toxicol 1996;34:959-62.
- [14] Diré GF, Lima EAC, Pereira MJS, Oliveira MBN, Moreno SRF, Mattos DMM, Jales RL, Bernardo-Filho M. Effect of a chayotte (*Sechium edule*) extract on the labeling of red blood cells and plasma proteins with technetium-99m: *in vitro* and *in vivo* studies. Cell Mol Biol 2002;48:751-5.

- [15] Felzenszwalb I, De Mattos JCP, Bernardo-Filho M, Caldeira-de-Araújo A. Shark cartilage-containing preparation: protection against reactive oxygen species. Food Chem Toxicol 1998;36:1079-84.
- [16] Gleason MN, Gosselin RE, Hodge CH, Smith RP. Clinical Toxicology of Commercial Products (Acue Poisoning), Williams and Wilkins, Baltimore, 1969.
- [17] Gordon EA. The antihypertensive effects of the Jamaican Cho-cho. West Indian Med J 2000;1:27-31.
- [18] Gutteridge JMC, Quilan GJ, Clarke I, Hallinwell B. Aluminium salts accelerate peroxidation of membrane lipids stimulated by iron salts. Biochem Biophys Acta 1985;835:441-7.
- [19] Leite JR, Seabra M. de L, Maluf E, Assolant K, Suchecki D, Tufik S, Klepacz S, Calil HM, Carlini EA. Pharmacology of lemongrass (*Cymbopogon citratus* Stapf). Assessment of eventual toxic, hypnotic and anxiolytic effects on humans. J Ethnopharmacol 1986;17:75-83.
- [20] Hallas LE, Cooney JJ. Tin and tin-resistante microorganisms in Chesapeake Bay. Appl Environ. Microb 1981;41:466-71.
- [21] Halliwell B. Free radicals and antioxidants: a personal view. Nutr Rev 1994;52:256-8.
- [22] Hladik III WB, Saha GB, Study KT. Essentials of Nuclear Medicine Science. Williams and Wilkins, Baltimore, London, 1987.
- [23] Jensen LP, Lai AR. Chayote (Sechium edule) causing hypokalemia in pregnancy. Am J Obstet Gynecol 1986;5:1048-9.
- [24] Lima EAC, Diré G, Mattos DMM, Oliveira MN, Mattos JCP, Dantas FJS, Caldeira-de-Araújo A, Bernardo-Filho M. Effect of the leaf extract from cauliflower (Brassica oleracea L.Var.Botrytis) on the biodistribution of the radiopharmaceutical sodium pertechnetate in mice and on the electrophoretic mobility of plasmid pUC 9.1 DNA. J Labelled Cpd Radiopharm 2001;44:642-4.
- [25] Lima EAC, Diré G, Mattos DMM, Freitas RS, Gomes ML, Oliveira MBN, Faria MVC, Jales RL, Bernardo-Filho M. Effect of an extract of cauliflower (leaf) on the labeling of blood elements with technetium-99m and on the survival of Escherichia coli AB1157 submitted to the treatment with stannous chloride. Food Chem Toxicol 2002;40:919-23.
- [26] Luria SE, Buroous JW. Hybridization between *E. coli* and *Shigella*. J Bacteriol 1957;74:461-76.
- [27] Mattos JCP, Dantas FJS, Bezerra RJAC, Bernardo-Filho M, Cabral-Neto JB, Leitão CLAC, Caldeira-de-Araújo A. Damage induced by stannous chloride in plasmid DNA. Toxicol Lett 2000;116:159-63.
- [28] McLean JRN, Blakey DH, Douglas GR, Kaplan JR. The effect of stannous and stannic (tin) chloride on DNA in

Chinese hamster ovary cells. Mutat Res 1983;119:195-201.

- [29] Melo SF, Soares SF, Costa RF, Silva CR, Oliveira MBN, Bezerra RJAC, Caldeira-de-Araújo A, Bernardo-Filho M. Effect of the *Cymbopogon citratus, Maytenus ilicifolia and Baccharis genistelloides* extracts against the stannous chloride oxidative damage in *Escherichia coli*. Mutat Res 2001;496:33-8.
- [30] Moro CO, Basile G. Obesity and medicinal plants. Fitoterapia 2000;71:S73-S82.
- [31] Oliver P, Marzin D. Study of genotoxic potential of 48 inorganic derivates with the SOS chromotest. Mut Res 1987;189:263-9.
- [32] Rader JI. Anti-nutritive effects of dietary tin. Adv Exp Med Biol 1991;289:509-24.
- [33] Reiniger IW, Silva CR, Felzenszwalb I, Mattos JCP, Oliveira JF, Dantas FJS, Bezerra AAC, Bernardo-Filho M. Boldine action against the stannous chloride effect. J Ethnopharmacol 1999;68:345-8.

- [34] Sallé JLO. Totum Em Fitoterapia. Robe Editorial, São Paulo, 1996:237.
- [35] Silva CR, Oliveira MBN, Melo SF, Dantas FJS, Mattos JCP, Bezerra RJAC, Caldeira-de-Araujo A, Duatti A, Bernardo-Filho M. Biological effects of stannous chloride, a substance that can produce stimulation or depression of the central nervous system. Brain Res Bulletin 2002;59:213-6.
- [36] Singh I. Introduction of reverse mutation and mitotic gene conversion by some metal compounds in *Saccharomyces cerevisae*. Mutat Res 1983;117:149-52.
- [37] Stohs SJ, Bagchi D. Oxidative mechanisms in the toxicity of metal ions. Free Radic Biol Med 1995;18:321-36.
- [38] Triphaty NK, Wurgler FE, Frei H. Genetic toxicity of six carcinogens and six non carcinogens in the *Drosophila* wing spot test. Mut Res 1990;242:169-80.
- [39] Wood JM. Biological cycles for toxic elements in the environment. Science 1974;183:1049-52.