Storage Stability and Quantitative Determination of Metalaxyl and Its Metabolite 2,6-Dimethylaniline in Wettable Powder (WP) Formulation

Nasr Sobhy Khalil⁽¹⁾ and Mohamed, A. A. Shaymaa⁽²⁾

⁽¹⁾ Pesticides Analysis Research Department, Central Agricultural Pesticides Laboratory, Agricultural Research

Center

⁽²⁾ Pesticides Formulation Research Department, Central Agricultural Pesticides Laboratory, Agricultural Research

Center

E-mail: <u>nasr_khalil@yahoo.com</u>

Abstract: A sample of Metalaxyl 35% WP was supplied by Central Agricultural Pesticides laboratory from a local company and the country of origin of the sample is Jordan. The physical and chemical properties of the Metalaxyl 35% WP samples were examined initially, cold stored at 0 °C \pm 1 for 7 days in the refrigerator and stored in glass bottles in the oven at 54°C \pm 2 °C for 3, 7, 14 and 28 days. The physical properties were carried out as follows Alkalinity, Density, Tapped density, pH and Wettability and the physical properties of their spray solutions under recommended dose in soft and hard were carried out as follows Persistent foam, Suspensibility, Conductivity, Surface tension and Viscosity. Generally, increasing time of hot storage for long time as 14 and 28 days at 54 °C may causes damage of the samples nature which may have diverse effects on their fungicidal activity against their targets. The storage variation causes basisty variation referred to pH, conductivity and salinity may causes phytotoxicity for plants during using their spray solutions. Finally, the Metalaxyl content decreased with the time of storage and still within the limit till 21 days of storage at 54 ± 2 °C. While after 28 days of storage the Metalaxyl content decreased till become out of the comply limit. On the other hand, the amount of metabolite 2,6-dimethylaniline increases with with the time of storage increase which may be reflected on the toxicity and biological activity of the sample.

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Key words: Metalaxyl 35% WP, storage conditions, physical and chemical properties.

1. Introduction

Metalaxyl (methyl N-(2-methoxyacetyl)-N-(2,6xylyl)-DL-alaninate) is a systemic fungicide used to control plant diseases caused by Oomycete fungi of the order PeronosporalesSchwinn et al., (1977) and Houseworth, (1987). Its formulations include granules, wettable powders, dusts, and emulsifiable concentrates. Application may be by foliar or soil incorporation, surface spraying (broadcast or band), drenching, and seed treatment. Metalaxyl registered products either contain metalaxyl as the sole active ingredient or are combined with other active ingredients (e.g., captan, mancozeb, copper compounds, carboxin). It has low mammalian toxicity and is regarded as safe for general use. However combination of trace impurities present in technical material arisingeither during synthesis orstorage, may lead to markedly deferent toxicities than would be expected from the toxicities of the individual components.

Adequate analytical methods of TLC, GLC, HPLC, MS, and other techniques are available for identification and determination of metalaxyl. In this work an analytical method was standardized for the estimation and determination of metalaxyl and its impurities (2,6-dimethylaniline) in a formulation of metalaxyl 35% WP in different storage periods at $54 \pm 2^{\circ}$ C. This work is concerned with the identification of the impurities, which are commonly present or may be developed upon storage and identified by FAO in formulation metalaxylwattable powder (WP).

Determination was by gas chromatography using flame ionization detector (FID). The identification of compound was based on areas and retention times by using external standard. The method showed decrease in metalaxyl content with the time of storage while its metabolite 2,6-dimethylaniline content increased.

2. Materials andMethods

1. Fungicide:

2. PhysicalParameters:

The physical properties of the Metalaxyl 35% WP samples under different storage condition were carried out according to CIPAC handbook methods as follows Alkalinity (MT 191), Density (MT 186), pH (MT 75.3) and Wettability (MT 53.3.3) **Dobrat and Martijn, 1995** where Tapped density (ASTM B-527) was carried out according to **American Society for**

Testing and Materials, 2006. Also the physical properties of their spray solutions under recommended dose in soft and hard were carried out as follows; Persistent foam (MT 47.2), Suspensibility (MT-54) and Conductivity(MT

32) (**Dobrat and Martijn, 1995**). Surface tension and Viscosity were carried out according to American Society of Testing and Materials (ASTM D-1331, 2014 and D-2196, 2015).

Fungicide:

Common Name:	Metalaxyl (fungicide)
Chemical Structure	$\begin{array}{c} O & CH_3 \\ CH_3O & C & CH \\ CH_2 & N & CO_2CH_3 \\ CH_3 & CH_3 \end{array}$
IUPAC name	methyl <i>N</i> -(methoxyacetyl)- <i>N</i> -(2,6-xylyl)-DL-alaninate; methyl 2-{[(2,6-dimethylphenyl)methoxyacetyl]amino}propionate
Chemical Name:	methyl N-(2,6-dimethylphenyl)-N-(methoxyacetyl)-DL-alaninate
CAS Number:	[57837-19-1]
Molecular Weight:	279.3
Molecular Structure:	C ₁₅ H ₂₁ NO ₄

3. ChemicalParameters:

A sample of Metalaxyl 35% WP was supplied by Central Agricultural Pesticides laboratory (Pesticide Analysis Research Department). It was provided by a local company and the country of origin of the sample is Jordan. The physical and chemical properties of the Metalaxyl 35% WP samples were examined initially, cold stored at 0 °C \pm 1 for 7 days in the refrigerator and stored in glass bottles in the oven at 54°C \pm 2 °C for 3, 7, 14 and 28 days according to WHO and FAO specifications (1988). According to these specifications the determined average active ingredient after 14 days content must not be lower than 97% relative to the determined average content found before storage noting that the samples of the product taken before and after the storage stability test should be analyzed together after the test to reduce the analyticalerror.

4. Analyticalprocedures:

The method used for the determination of metalaxyl and its metabolite (2,6-dimethylaniline) in a formulation sample of metalaxyl 35% WP by GC (FID) on a capillary column using external standard.

4.1 Apparatus:

Gas chromatography- GC-system Agilent technologies 7890A instrument equipped Flame ionization detector (FID). A fused silica capillary column (30 m length, 0.53 mm-i.d., coated with methyl silicon gum HP-1, with 2.65 lm film thickness) was used. Nitrogen was used as a carrier gas with a flow rate of 20 ml. The injector (split less capillary

inlet system, injection volume 1 µl) and detector temperatures were 250°C and 300°C, respectively.

Oven temperature was programmed from 70°C to 240°C (3 mins) at a rate of 25° C / min for the determination of impurities of metalaxyl (2,6-dimethylaniline). While Oven temperature was programmed from 150°C to 270°C (3 mins) at a rate of 25° C / min for the determination of impurities of metalaxyl. Gas and corresponding flowrates: Nitrogen (carrier) 25 ml per min.

-	-
Hydrogen	40 ml permin.

400 ml	permin

4.2 Reagents

Air

- A. Acetone HPLC grade
- B. Metalaxyl analytical standard (purity of 99%) wasobtained from the Central laboratory of Agriculturalpesticides.
- C. 2,6-dimethylaniline (purity of 99% were obtained fromSigma-Aldrich).

4.3 Preparation of standardsolution

Weigh accurately 0.01 gm of 100 % metalaxyl reference standard (0.0101 gm of 99% metalaxyl reference standard) into 25 ml volumetric flask and add 25 ml of acetone. Shake well to homogenize.

Repeat this step using 2,6-dimethylaniline reference standard.

4.4 Preparation samplesolution:

For metalaxyl determination, weigh accurately a quantity of the sample (0.0286 gmmetalaxyl 35% WP) equivalent to 0.01 gm of 100 % metalaxyl reference standard into a 25 ml volumetric flask and add 25 ml of acetone. Shake well to homogenize. For

2,6-dimethylaniline determination, weigh accurately 1 gmof the sample (metalaxyl 35% WP) into a suitable 1 volumetric flask, and complete to 25 ml by acetone. Shake well tohomogenize.

5.5 Estimation:

Inject 1 μ l from the reference standard solution into the GC followed by the sample note the peak area and the retention time of Metalaxyl stander and sample.

3. Results and Discussion

1. Physical Parameters:

1.1. The Metalaxyl 35% WPSamples

The data in table (1) illustrated the physicochemical properties; alkalinity, bulk density, tapped density, pH at 1% solution and Wettability of metalaxyl 35% WP under different storage conditions; zero time (initial), cold storage at 0 °C for 7 days and hot storage at 54 \pm 2 °C for 3, 7, 14 and 28 days; respectively.

Table	(1):	Physica	l Pro	perties (of Metalax	vl 35%	WP	under	Different	Storage	(Hot an	d Cold)	Conditions.
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T-mag of stome and	Initial	Cold storage	Hot storages at 54 ⁰ C for					
i ypes of storages	time 0	at 0 °C for 7	3 davs	7 davs	14 davs	28 days		
Free alkalinity(% of NAOH)	0.20	0.17	0.23	0.27	0.3	0.35		
CIPAC MT 191								
PH of 1% solution	9.5	8.79	9.5	9.9	10.3	10.7		
CIPAC MT 75.3								
Density (gm/cm ³)	0.588	0.631	0.583	0.575	0.559	0.531		
CIPAC MT 186								
Tapped Density (gm/cm ³)	0.769	0.851	0.746	0.714	0.694	0.641		
ASTM B-527								
Hausner Index (H.I)	1.37	1.34	1.28	1.24	1.24	1.21		
Compressibility factor (C.F)	28.5	25.8	21.3	19.5	19.4	17.1		
Wettability per second	Less than	Less than	Less	than	more than	n 2 minutes		
CIPAC MT 53.3.1	2	2	2					

1.1.1 Alkalinity(%):

The alkalinity of metalaxyl 35% WP samples under different storage conditions; initial, cold and hot storage for 3, 7, 14 and 28 days were 0.2, 0.17, 0.23, 0.27, 0.3 and 0.35 % as NaOH. The samples of initial, cold and hot storage for 3 and 7 are acceptable where the other samples of hot storage for 14 and 28 days extremely increased by about 50% and 75% of the initial samples which may reflect the break of their formulations.

1.1.2 PH of 1%solution:

The pH values of 1% solution of metalaxyl 35% WP samples were alkaline, their alkalinity increased from initial time to hot storage at different time intervals 3, 7, 14 and 28 days as follows; 9.5, 9.5, 9.9, 10.3 and 10.7, respectively where the later 2 samples may causedphytotoxicity on plants. The cold storage sample was less alkaline than the other samples where it value was 8.8 approximately.

1.1.3 BulkDensity:

The bulk density of Metalaxyl samples were decreased descendingly from initial to the hot storage at different time intervals 3, 7, 14 and 28 days and their values were 0.588, 0.583, 0.575, 0.559 and 0.531 gm/cm³ were the metalaxyl samples of cold storage was the highest value 0.631 gm/cm³.

1.1.4 Tapped density(cp):

Also the tapped density of metalaxyl samples take the same way for the density values where the tapped density decrease descendingly from initial (zero time) to the hot storage at different time intervals 3, 7, 14 and 28 days and their values were 0.769, 0.746, 0.714, 0.694 and 0.641 gm/cm³ which appeared that by increasing the time of hot storage the impacting of the samples increased that may be destroying the nature of wettable powder characters of its formulation.

On the other hand, Carr's compressibility factor (C.F %) and hausner index (HR), are measuring of the powder flow tendency. These two factors measured according to the equations

C.F = $[(\rho_{tap} - \rho_{bulk}) / \rho_{tap}] X100$

H.I. = ρ_{tap} / ρ_{bulk}

The data presented in table (1) showed the flow properties of the tested fungicide through different storage conditions. The samples of hot storages for 14 and 28 days, theirHausner indexes were lower than 1.25 and also their Compressibility factor were lower than 20.0% which indicates that the powder is poor flowability according to **Abdullah and Geldart** (**1999**) while, the other storage conditions were free flowing where their H.I higher than 1.25 and alsotheir C.F > 20.0% means which agree with **Schüssele and Bauer-Brandl (2003)**.

1.1.5 Wettability:

The Wettability nature of wettable powder formulation must take place not more than 2 minutes without swirling according to the FAO and WHO specification. According to this fact, the initial, cold and hot storage for 3 and 7 days metalaxyl 35% WP were acceptable as a wettable powder formulation where their wetting time were less than 2 minutes while the other samples of 14 and 28 days took more than 2 minutes which reflect the break of its formulation as a wettable powder.

Generally, the later study on Metalaxyl35% WP samples appeared that, increasing time of hot storage for long time as 14 and 28 days at 54 $^{\circ}$ C

which give indication for the stability of the sample during two years of manufacturing **Cowlyn** (1993). Also it may cause damage of the samples nature which may have diverse effects on their fungicidal activity against their targets.

1.2 Spray solution of Metalaxyl30% WPSamples:

The data presented in table (2) illustrated the physico-chemical properties; persistence foam, pH, conductivity, salinity, suspensibility, viscosity and surface tension of metalaxyl 35% WP samples spray solutions (soft and hard) water under different types of storage (initial time, cold storage at 0 °C for 7 days and hot storage at 54 °C for 3, 7, 14 and 28 days

 Table (2): Physical Properties for Spray Solutions of Metalaxyl 35% WP under Different Storage (Hot and Cold) Conditions.

	Initia		Cold sto	orage at	Hot storages at 54 ^o C for								
Test	l time		⁰ C for 7 days		3 days		7 Days		14 Days		28 Days		
	S	н	S	н	S	Н	S	н	S	н	S	Н	
Foam per ml	2	3	2	3	3	4	3	5	7	9	11	15	
PH	9.81	9.93	8.80	8.98	9.90	9.99	10.01	10.10	10.30	10.45	10.40	10.73	
Conductivity 🛛 mhos	210	710	200	860	240	750	250	790	310	850	350	910	
Salinity %o	0.2	0.7	0.2	0.8	0.2	0.7	0.2	0.7	0.3	0.8	0.3	0.9	
Suspensibility %	78	75	75	75	73	71	67	61	63	57	55	50	
Viscosity (cpois)	2.13	2.15	2.14	2.16	2.10	2.15	2.2	2.2	1.81	1.72	1.65	1.53	
Surface Tension Dyn/cm	2	27	2	5	26		26 26		30		32		

S= softwater

H=hardwater

1.2.1 FoamingTest:

The percentage of foam for spray solutions soft and hard water of metalaxyl samples at initial time, cold storage and hot storage for 3, 7 and 14 days were acceptable while it was not acceptable for sample hot stored for 28 days because its value exceed the 10%.

1.2.2 pH:

All hard water spray solution samples of metalaxyl under different storage conditions their pH values were greater than their soft water samples. The

pH range of spray solutions varied between 8.8 for soft water cold storage as minimum values where it was more alkaline for hard water of 28 days. Hot storage samples and its value were 10.73. The alkalinity of soft and hard water spray solutions for 14 and 28 days; where the highest value which reflects the effect of their hotstorage.

1.2.3 Conductivity:

Also, the conductivity of all hard water spray solutions was higher than that of soft water spray solutions at the same conditions. The conductivity of spray solution soft and hard increased from initial then followed by cold storage, hot storage for 3, 7 and 14 days where the conductivity of soft, hard water spray solutions for 28 days were the highest where the values were 350 and 910μ mhos. **1.2.4 Salinity:**

Also, the salinity as a percentage of thousands for hard water spray solution is greater than of soft waterof metalaxyl samples of the same conditions. The salinity of soft water solutions were 0.2 ‰ as a lower values for initial, cold storage and hot storage samples for 3 and 7 days while the salinity of hard water spray solution were the highest for 28 days hot storage at $54 \pm 2^{\circ}$ C.

1.2.5 Suspensibility:

Also the salinity, the suspensibility of hard water spray solution is greater than of soft water of metalaxyl sample of the same conditions. The salinity for soft water solution of metalaxyl were 0.2 % as a lower values for initial time, cold storage and hot storage for 3, 7 at 54 \pm 2 °C were passed suspensibility test and their values were 78, 75, 75, 73 71, 67 and 65% respectively. Soft and hard water spray solutions for hot storage for 14 and 28 days were failed in suspensibility because their values were 60, 57, 55 and 50%, respectively.

1.2.6 Viscosity:

The viscosities of soft and hard water spray

solutions for metalaxyl samples at initial, cold and hot storage for 3 and 7 days varied in the range of 2.10 to 2.2 cp and their average values was 2.15 cp but the other viscosities values were decreased than this average by almost 25:40% of their values.

1.2.8 Surface Tension:

The surface tension of the spray solution soft and hard water for the four types of storage was approximately constant in the range of 26 dyne/cm² \pm 1 but the surface tension of the two later hot stored samples for 14 and 28 days were increased by 15 and 23 %, respectively.

Finally, the initial, cold and hot storage for 3 and 7 days at 54 \pm 2 °C did not affecton the Metalaxyl35% WP samples. Physical properties like alkalinity, bulk density, tapped density, pH and wettability were not affected where the others were affected by the long timeof storage and distorted the WP formulation nature which reflects on the physicochemical properties of their spray solutions in both soft and hard water. They were failed in persistence foam, suspensibility and make decrease in viscosity of their spray solutions which decreased their adhering on the surface of the leaf. Also, their surface tension increased which increasing the drift of spray solutions drops from the surface of plant leaves Moustafa, et al. (1990) and Tawfik, Mona and El-Sisi (1987). The storage variation causes basisty variation referred to pH El-Kady (2007) and Hussein, et al. (2009); conductivity and salinity may cause phytotoxicity for plants during using their spray solutions.

2. Chemicalanalysis:

The research around pesticides is a growing area in the field of analytical chemistry perhaps due to the fact that every year new pesticides are synthesized in order to achieve more effective products with a minor application rate (Sekhon, 2009). Metalaxyl [(R,S) methyl-N-(2-methoxyacetyl)-N-(2,6-xylyl)-dl-

alaninate is an acylamine fungicides (included into the amide group), being the most widely known member of this group. This fungicide, synthesized in 1977, is widely used to control plant diseases caused by pathogens of the Oomycota division, in particular, against *Phytophthorainfestans* and *Phytiumultimum* **Fischer**, *et al.* (1982). The active ingredient of the commercially metalaxyl 35 % WP formulation used in this experiment is the S-enantiomer. Many research workers have studied the effect of metalaxyl as a fungicide but the effects of temperature and storage stability on pesticide breakdown and their effect on the activity of the pesticide have not generally been described quantitatively more enough.

Metalaxyl and its metabolites containing 2,6dimethylaniline have been determined by GLC with a flame ionization nitrogen detector Anon., (1995). Quantitative estimation of Metalaxyl and its metabolite 2,6- dimethylaniline impurity present in the samplewas done by comparing the area of known concentration of purified and known concentration standard with those in the samples. The values of areas, migration times and resolution were obtained Station using the Chem. software. The Metalaxylcontent differ from that declared (35 %) by more than the followingamounts.

Declared content Permitted tolerance

Up to 250 g/kg \pm 6% and above 250 up to 500 g/kg \pm 5% of the declared content.

Maximum amount of the metabolite 2.6dimethylaniline is 0.1% of the metalaxyl content (1g/kg) according to FAO specifications of metalaxylwettable powders. As the amount of the metabolites is very minor as shown from the specification, their determination requires analytical methods sensitive enough to be able to detect and quantify. 2,6- dimethylaniline is amino aromatic compound used as a chemical intermediate in the manufacture of pesticides and other products Ethyl corp, (1990) and Kuney, (1991). It is used for the synthesis of metalaxyl, and may be present in trace amount in the sample, which may further react with acid chloride the manufacturing process to give the methoxyacetylated impurity. According to FAO specification its amount should not exceed 0.1 % of the total metalaxyl content in the sample.

Injected solution	Retention time (min.)	Retention time (min.) Storage period		Metalaxyl content Percent by mass	
Metalaxyl St solution	4.275		5539.95020		
Metalaxyl sample solution	4.275	Before storage	5706.82666	36.05 %	
Metalaxyl sample solution	4.275	2 days	5549.79443	24.84.04	
	4.274	5 days	5478.30420	34.04 %	
	4.278	7 dava	5372.93994	24.14.0/	
Metalaxyl sample solution	4.279	7 days	5436.11719	54.14 %	
	4.279	14 darm	5301.20898	22 56 0/	
Metalaxyi sample solution	4.279	14 days	5323.53564	55.30 %	
Metalaxyl sample solution	4.283	21 days	5255.21045	33.37 %	
Metalaxyl sample solution	4.283	28 days	5048.1858	32.06 %	

Table 3: Shows the retention time, peak area and concentration of metalaxyl in the formulation sample at different periods of storage:

Table 4: Shows the retention time, peak area and concentration of metalaxyl metabolite 2,6dimethylaniline in the formulation sample at different periods ofstorage:

Injected solution	Retention time (RT) (min.)	Storage period	Peak area	2,6-dimethylaniline content (gm/kg) formulation
2,6-dimethylaniline St solution	3.692		7882.08789	
2,6-dimethylaniline sample solution	3.690	Before storage	4.48336	0.0057
2,6-dimethylaniline sample solution	3.691	3 days	4.29857	0.0055
2,6-dimethylaniline sample solution	3.691	7 days	5.18836	0.0066
2,6-dimethylaniline sample solution	3.691	14 days	5.26285	0.0067
2,6-dimethylaniline sample solution	3.691	21 days	5.73877	0.0073
2,6-dimethylaniline sample solution	3.691	28 days	6.32968	0.008

Table 5: Shows the retention time, peak area and conce	entration of metalaxyl metabolite 2,6-
dimethylaniline in the formulation sample at different	periods ofstorage:

Injected solution	Retention time (RT) (min.)	Storage period	Peak area	2,6-dimethylaniline content (gm/kg) of Metalaxyl content
2,6-dimethylaniline St solution	3.692		7882.08789	
2,6-dimethylaniline sample solution	3.690	Before storage	4.48336	0.0158
2,6-dimethylaniline sample solution	3.691	3 days	4.29857	0.0157
2,6-dimethylaniline sample solution	3.691	7 days	5.18836	0.0193
2,6-dimethylaniline sample solution	3.691	14 days	5.26285	0.0199
2,6-dimethylaniline sample solution	3.691	21 days	5.73877	0.0218
2,6-dimethylaniline sample solution	3.691	28 days	6.32968	0.0250

The results of tables (4) and (5) show that 2,6dimethylaniline amount increases with the time of storage but within the limit of specification. During the first 14 days its concentration were scarcely increased, which agree with **Oscar Gorodo (2010)**. While table (3) shows that metalaxylcontent decreases with the time of storage and still within the limit and comply with FAO specification till 21 days of storage at 54 ± 2 °C. While after 28 days of storage the metalaxyl content decreased till become out of the comply limit which may be reflected on the toxicity and biological activity of thesample.

References

- 1. Abdullah, E.C. and Geldart, D (1999). The useof bulk density measurements as flowability indicators, Powder Technol. 102,151–165.
- 2. American Society of Testing and Materials (ASTM-2006). Standard Test

Methodof Tapped density.B-527.

- American Society of Testing and Materials (ASTM-2014). Standard Test Methodof surface TensionD-1331.
- 4. American Society of Testing and Materials (ASTM-2015). Standard Test Method viscosityD-2196.
- 5. Anon., 1995. Information on GAP of Metalaxylfor the 1995 JMPR by Agriculture and Agri-Food Canada, Food Production and Inspection Branch, February 23,1995.
- Anonymous 1988. FAO specifications, Pesticides Wettable powder, Storage Stability 54 ± 1 °C for 14 days p.19-20. CIPAC E, p.128, 365/ WP/M/3- metalaxyldetermination CIPAC E, p.128, 365/ WP/M/4- 2,6dimethylanilinedetermination
- 7. Cowlyn, T. C. (1993). Glyphosate ammonium salt determination of physicochemical properties. Publishing

Monsanto. Report no. 93/MON032/0343 (1995)Germany.

- Dobrat, W. andMartijn A. (1995). CIPAC hand book, Volumes F. MT 32; 36.2; 47.2; 53.3.3; 54; 75.3; and186.
- 9. Dobrat, W. andMartijn A. (2001). CIPAC hand book, Volumes L. MT191.
- 10. El-Kady, A. M. A. (2007). Correlation between high density/polyethylene and physicochemical properties packages of some pesticides in liquid Formulation tropical storage conditions. under J. Agric. Sci. Mansoura Univ., 32(12):10447-10453.
- 11. Ethyl corp (1990). OrthoalkylatedAnilines-2,6dimethylaniline (DMA), Baton Rouge,LA.
- 12. Fischer, D.J. andHayes, A. L (1982). Pestic. Sci.13-330.
- 13. Houseworth, L.D., (1987). Excerpts from the new products and services from industry. Plant Dis. 71, 286 -288.
- 14. Hussein, M. A; El-Kady, A. M. A.; Farag, E. M. A. andH. M. El-Torkey(2009) Polyethylene packagesof liquid pesticide formulations as an important factor in physico-chemical properties of stored pesticides under tropical storage. Egypt. J. of Appl. Sci., 24(3):326-337.
- 15. Manual on development and use of FAO and WHO specifications for pesticides, (2002). 1stedition. Pp. 121–123.
- 16. Manual on development and use of FAO

and WHO specifications forPesticide Specifications, (2006). prepared by the FAO / WHO Joint Meeting on Pesticide Specifications (JMPS) P.36

- 17. Manual on development and use of FAO and WHO specifications for pesticides, (2006). 2^{nd} edition. Pp. 125 –127.
- Moustafa, O. K.; Z. M. El-Attal and A. G. El-Sisi (1990). Influence physicochemical properties of water on the performance and efficiency some insecticides. Agric.Res.Rev., Egypt, P.127 –133.F.
- 19. Oscar Gordo, Juan Jpse Sanz and Morj M. Lobo (2010). Storage stable pesticides formulations containing azadirachtin. J. of Insect Science, 10:34.
- Schüssele, A and Bauer-Brandl, A (2003). Note on the measurement of flowability according to the European Pharmacopoeia, Int. J. Pharm. 257,301–304.
- Schwinn, F.J., Staub, T., Urech, P.A., (1977). A newtype of fungicide against diseases caused by oomycetes. Meded. Fac. Landbouwwet, Rijksuniv. Gent. 42, 1181 -1188.
- 22. Sekhon, B.S. J. Pestic. Sci. 34 (2009)1.
- Tawfik, Mona H. and A. G. El-Sisi (1987). Persistent of foliarfertilizers on the physical properties and insecticidal activityof locally spray oils against the scale insect *Parlatoriaziziphus*(Lucas) Naf. Conference of Pests & Diseases of Vegetables & Orchard, Ismaillia, Egypt, Pp.376.

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